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O Laurent Lavoisier

Chemistry Quiz: Trivia Questions and Answers

1. How atoms and molecules interact via chemical bonds?
 An interaction between two atoms, which involves the sharing of one or more electrons to help each atom satisfy the octet rule.
 A link or force between neighboring atoms in a molecule or compound.
 An attraction between two ions used to create an ionic compound.
2. Who have developed a law of conservation of mass?
O Antoine Laurent Lavoisier

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3. The particular number of protons in the nuclei of its atom is known as
O Mass numbe
O Atomic number
O Element
4. What is the name of standard presentation of the chemical elements?
O Elements Order
O Periodic table
Periodic trends
5. What is molecule?
O Valence electrons
O Electrically neutral units
 Atoms bound together by covalent bonds
6. What is the name of the things that do not split up during acid-base reactions?
O hydroxide (OH-) and phosphate (PO43-).
O Polyatomic ions
O Na+ and Cl- ions
7. Which energy is an attribute of a substance as a consequence?
O Chemical transformation
O Atomic, molecular or aggregate structure
O Increase or decrease of energy
8. What will be the chemical reaction in the formation or dissociation of molecules?
Breaking apart to form two or more smaller molecules
Rearrangement of atoms within molecules
Rearrangement of atoms across molecules
9. Which methods can be used to measure the Acid strength?

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O Study of the elementary particles

0	Arrhenius definition of acidity and Bronsted–Lowry definition
0	Acid dissociation constant (Ka)
0	Hydronium ion concentration
10	. What is the current model of atomic structure?
0	The Quantum mechanical model
0	The interactions, reactions and transformations



Chemistry Quiz: Trivia Questions and Answers Right answers

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- 2. Who have developed a law of conservation of mass? Antoine Laurent Lavoisier
- 3. The particular number of protons in the nuclei of its atom is known as Atomic number
- 4. What is the name of standard presentation of the chemical elements? Periodic table
- 5. What is molecule? Atoms bound together by covalent bonds
- 6. What is the name of the things that do not split up during acid-base reactions? hydroxide (OH–) and phosphate (PO43–).
- 7. Which energy is an attribute of a substance as a consequence? Atomic, molecular or aggregate structure
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- 9. Which methods can be used to measure the Acid strength? Arrhenius definition of acidity and Bronsted–Lowry definition
- 10. What is the current model of atomic structure? The Quantum mechanical model